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Total Synthesis of Astaxanthin and Hydroxyechinenone

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ASTAXANTHIN (I, R = d) occurs in *Crustacea* and other animals;^{1,2} together with its esters, it constitutes the principal prosthetic group of the carotenoproteins.^{1,3,4} Attempts to isolate astaxanthin frequently yield the corresponding diosphenol, astacene (I, R = b).¹ The latter may readily be obtained by autoxidation of canthaxanthin (I, R = a),^{5,6} for which a number of syntheses have been developed.⁷ We now report the conversion of astacene into astaxanthin by reduction with potassium borohydride, and oxidation of the resulting mixture of tetraols (I, R = c) with manganese dioxide or, preferably, 2,3-dichloro-5,6-dicyanoquinone.

The product, which exhibited a molecular ion† corresponding to $C_{40}H_{52}O_4$, was identified by direct comparison with an authentic specimen‡ from the common lobster, *Homarus gammarus* L. (mixed thin-layer chromatograms on Kieselgel H). Both had λ_{max} (CS₂) 503 m μ , λ_{max} (CHCl₃) 485 m μ , λ_{max} (MeOH) 472 m μ , ν_{max} (CHCl₃) 3520 and 1660 cm.⁻¹, τ (CDCl₃)* 8.74, 8.06, and 8.02, and yielded astacene on autoxidation.

Synthetic astaxanthin combined[‡] with the

† Precision mass spectrometry on an A.E.I. MS9 spectrometer with sample directly inserted into source (Dr. E. S. Waight).

* Methyl bands only.

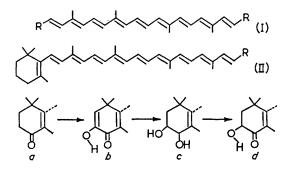
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appropriate apoprotein from the lobster carapace to give the characteristic blue colour, λ_{max} (phosphate buffer, pH 7) 630 m μ , of α crustacvanin.4

Autoxidation of echinenone (II, R = a) gave 3-oxoechinenone (II, R = b).⁶ Reduction of the latter with potassium borohydride, and oxidation of the resulting glycols (II, R = c) with 2,3dichloro-5,6-dicyanoquinone or, preferably, acetone and aluminium t-butoxide, gave 3-hydroxyechinenone (II, R = d), m.p. (evac. capillary) 156-157°. This exhibited the expected spectral properties, λ_{max} (C₆H₆) 472 m μ , λ_{max} (CHCl₃) 472 m μ , λ_{max} (EtOH) 460 m μ , λ_{max} (petrol) 457 m μ , ν_{max} (CCl₄) 3520 and 1665 cm⁻¹, τ (CDCl₃)* 8.96, 8.74, 8.28, 8.06, and 8.02, and a molecular ion^{\dagger} corresponding to C₄₀H₅₄O₂.

A pigment in Adonis annua has been formulated

as (II, R = d);⁸ this structure has now been confirmed by mixed chromatograms¶ of the derived diosphenol with 3-oxoechinenone.



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